

ACHEMA, FRANKFURT, JUNE 10-14, 2024

e-CODUCT:

ACHEMA2024



# Fast-Response electrically heated catalytic reactor technology for CO<sub>2</sub> reDUCTion

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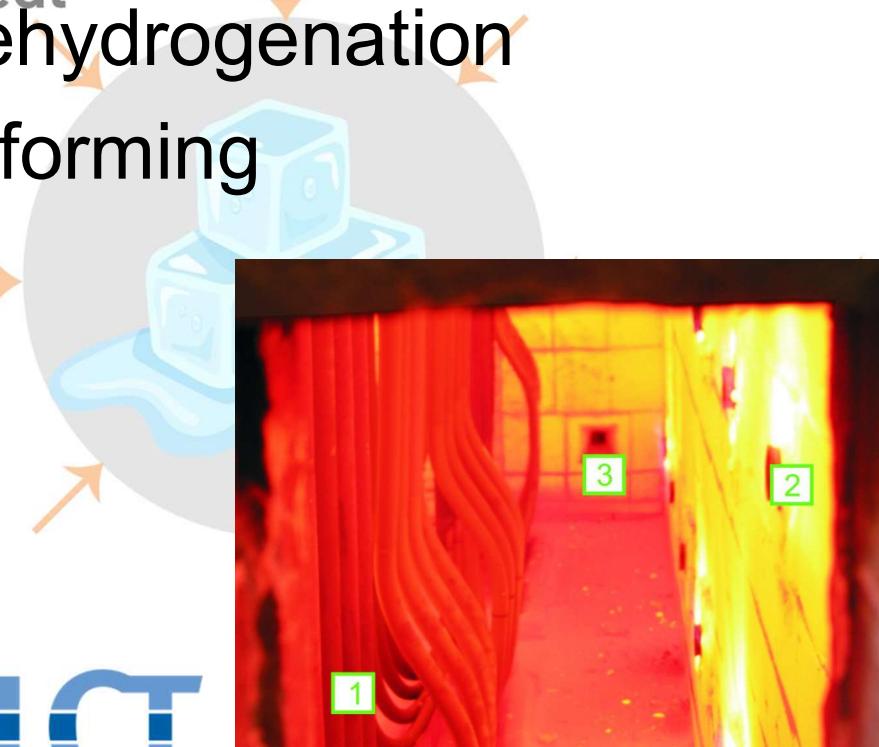
<sup>1</sup>Laboratory for Chemical Technology (LCT), Ghent University, Technologiepark 125, 9052, Ghent , Belgium

<sup>2</sup>TotalEnergies OneTech, Feluy, Belgium



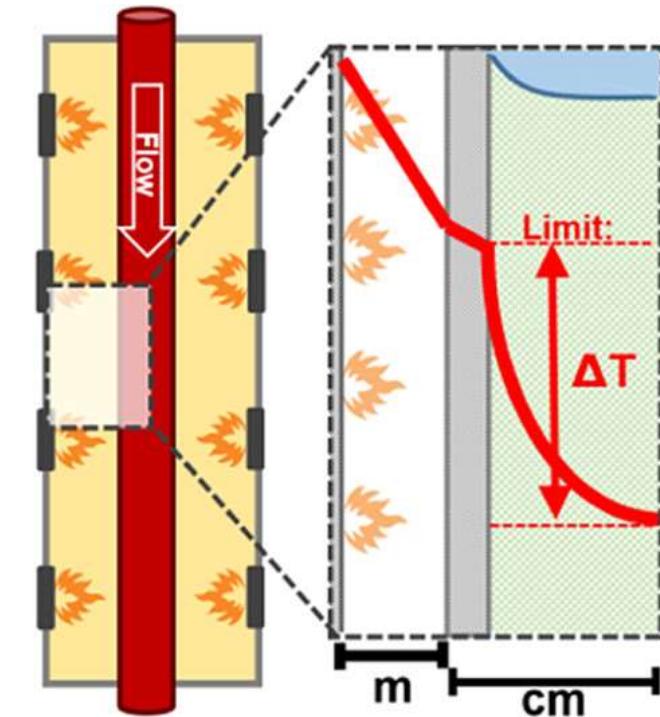
# heat requirements in chemical reactions

- endothermic reactions
  - thermal cracking
  - dehydrogenation
  - reforming
  - ...
- exothermic reactions
  - methanol synthesis
  - Fischer Tropsch
  - oxidation
  - ...



# (strongly) endothermic reactions

- heat transport focused reactor design
  - narrow tubes
  - fired furnace
- pronounced temperature gradients



# challenges/objectives

- enhancing heat transfer efficiency
- faster response to temperature changes

# outline

- introduction
- efficient heating
  - electrification
  - ElectroThermal Fluidized Bed reactor (ETFB)
- e-CODUCT
  - opportunities for simultaneous CO<sub>2</sub> and H<sub>2</sub>S abatement
  - powered by renewable energy
- conclusions & perspectives

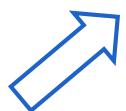
# how can we enhance heat transfer?



heat containment

electrification

# how can we do even better?



microwave



induction

heating from  
the inside

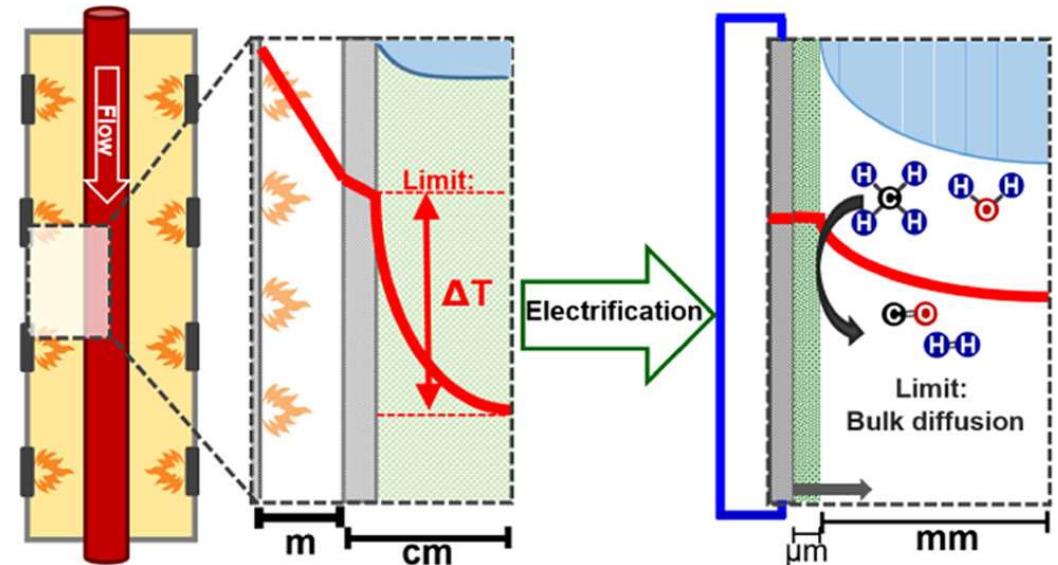


# reactor electrification

electrical heating -> overcoming limitations of combustion

advantageous in terms of:

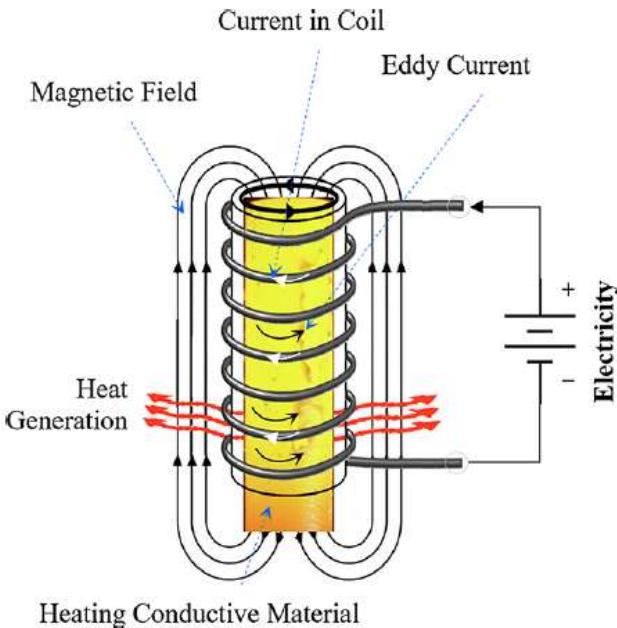
- energy efficiency
- process control
- safety and maintenance
- rapid heating
- ...



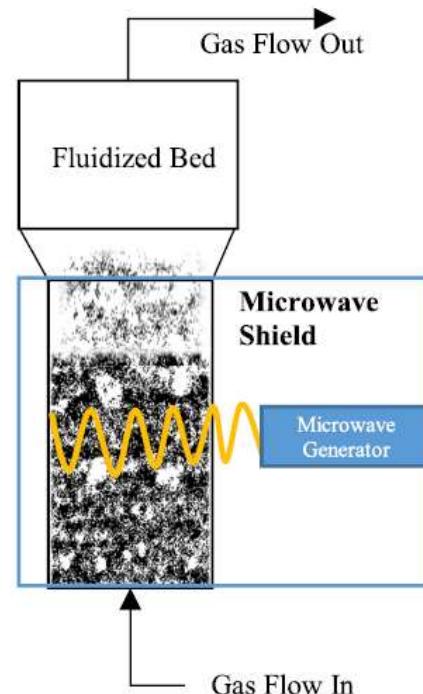
Wismann, et al. *Ind. Eng. Chem. Res.* 58 (2019) 23380

# electrical heating

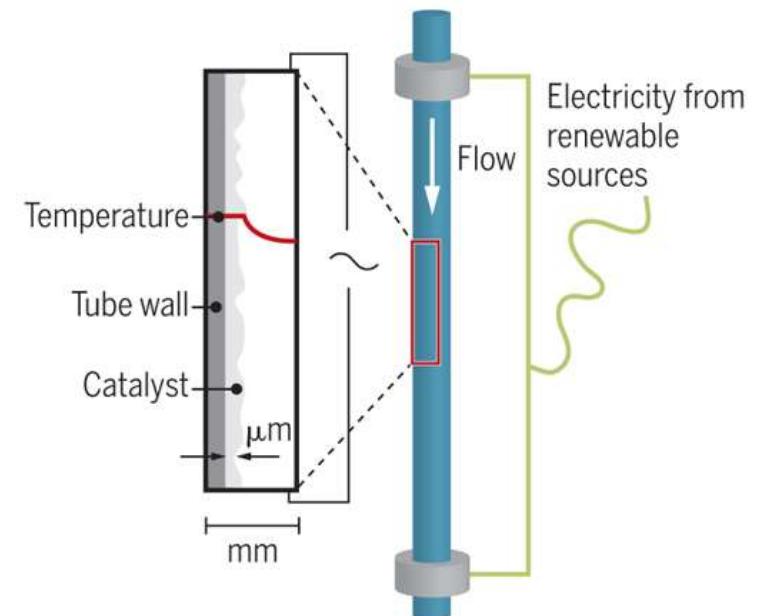
## induction



## microwave

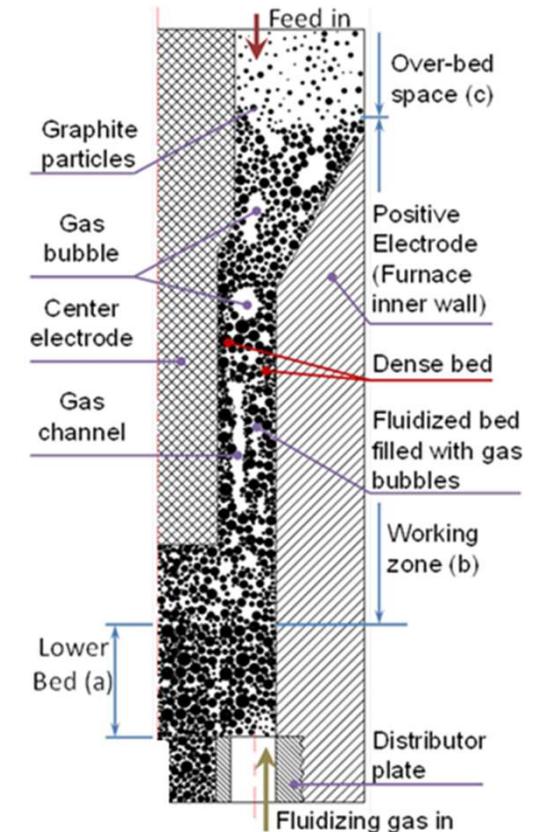


## Joule



# ElectroThermal Fluidized Bed reactor (ETFB)

- combination:
  - fluidization
  - Joule heating
- compared to conventional fluidized beds
  - better control over bed temperature
  - highly energy efficient
  - rapid and uniform heating



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# e-CODUCT: rationale

fast-response electrically heated catalytic reactor technology for CO<sub>2</sub> reduction

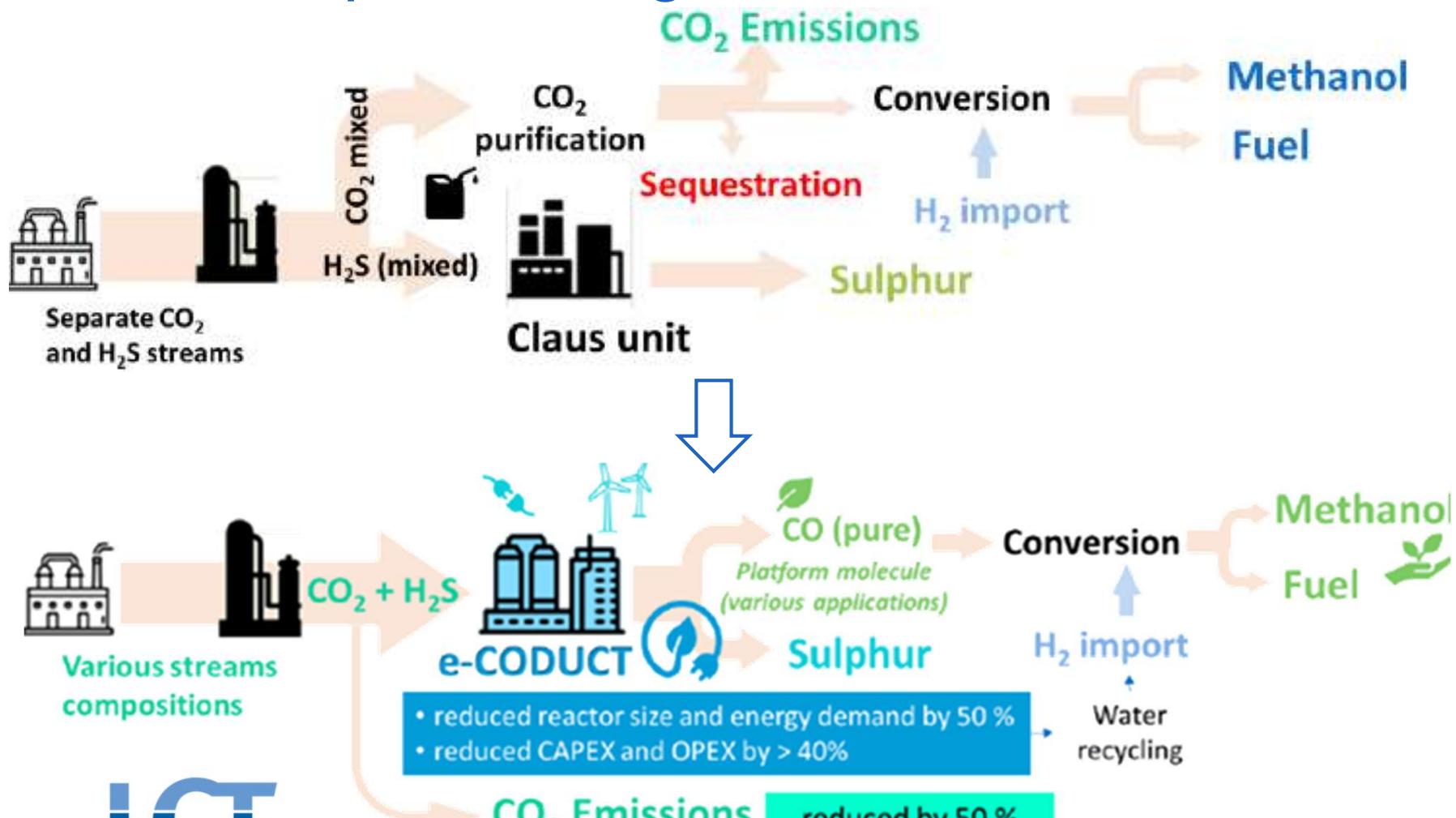


why?

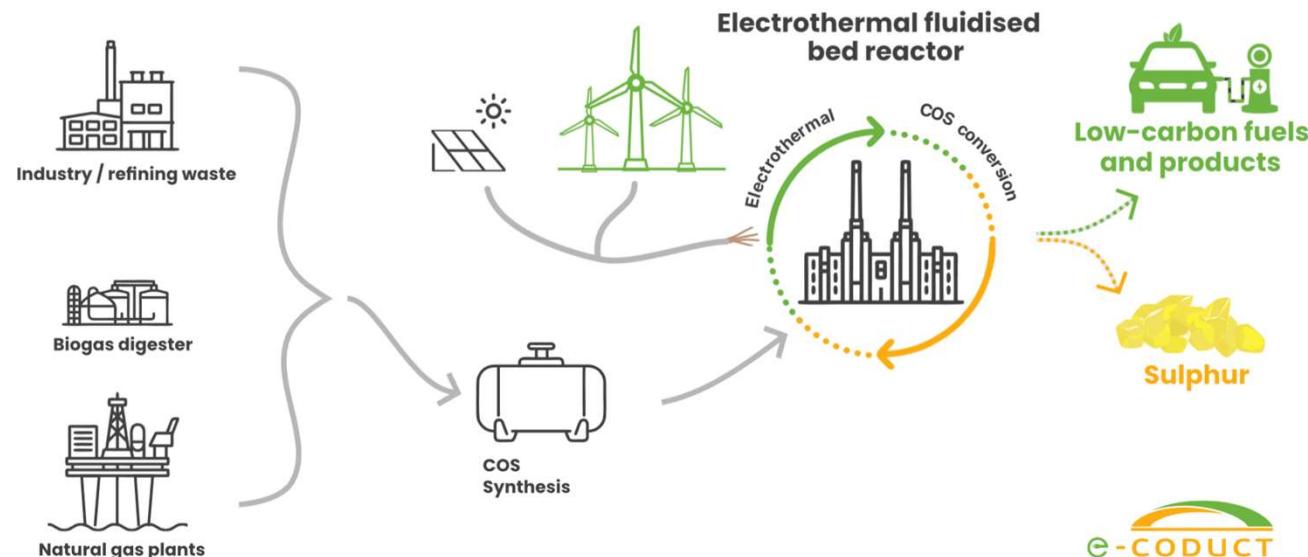
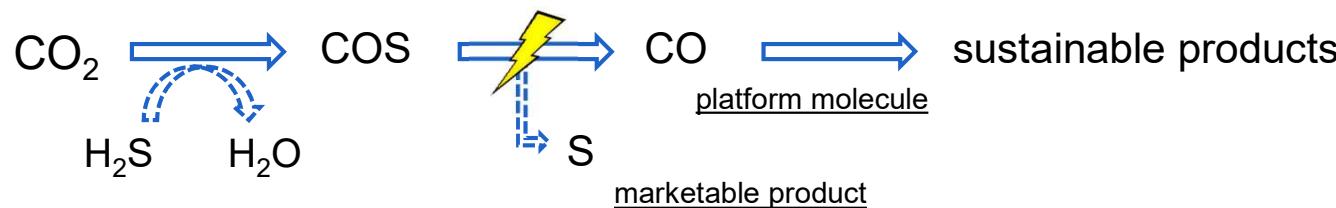
- current CO<sub>2</sub> reduction technologies require highly pure streams
- no existing technologies for simultaneous CO<sub>2</sub> and H<sub>2</sub>S reduction
- making more feedstock sources available



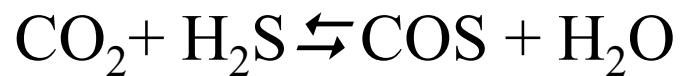
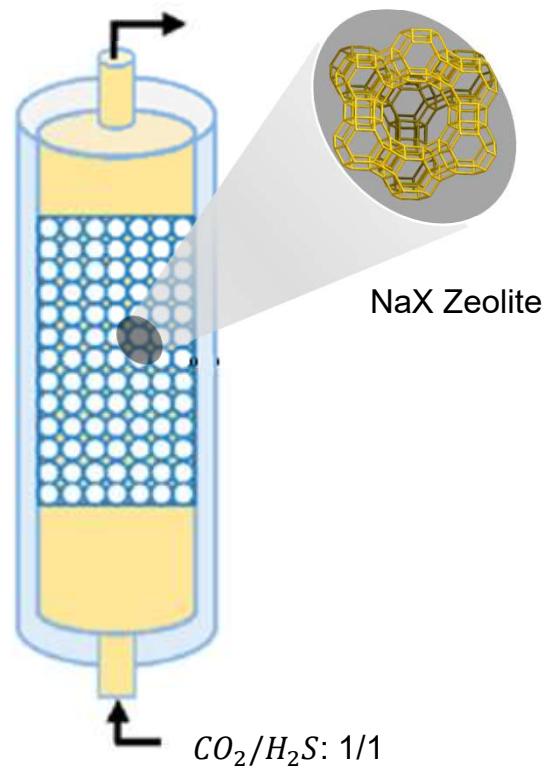
# e-CODUCT: positioning



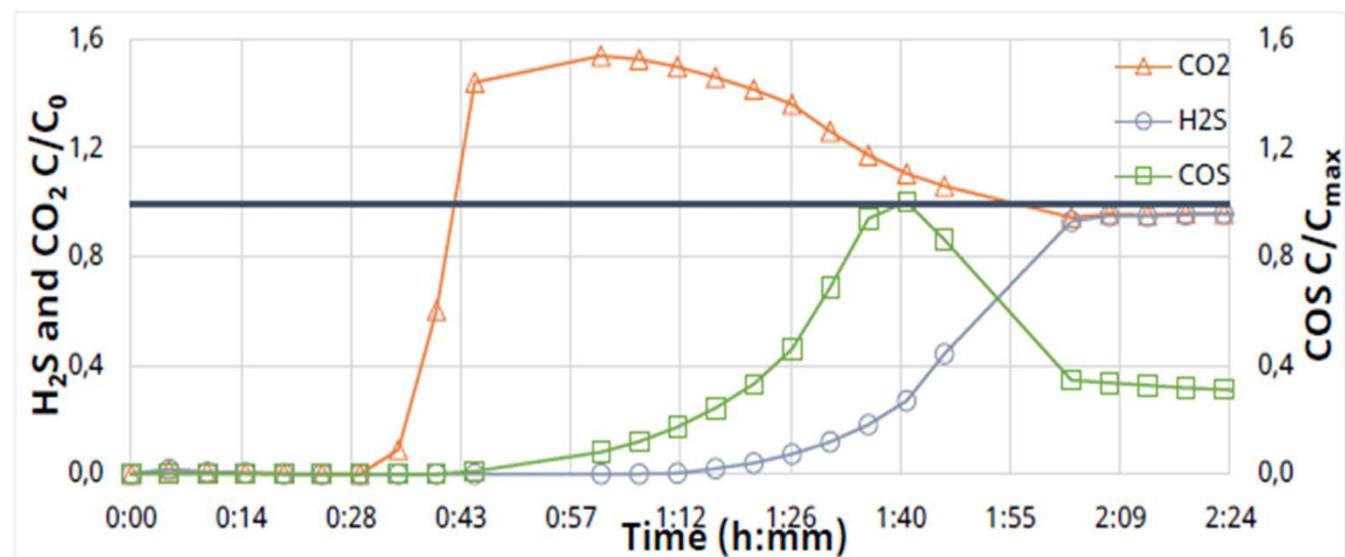
# e-CODUCT: process lay-out



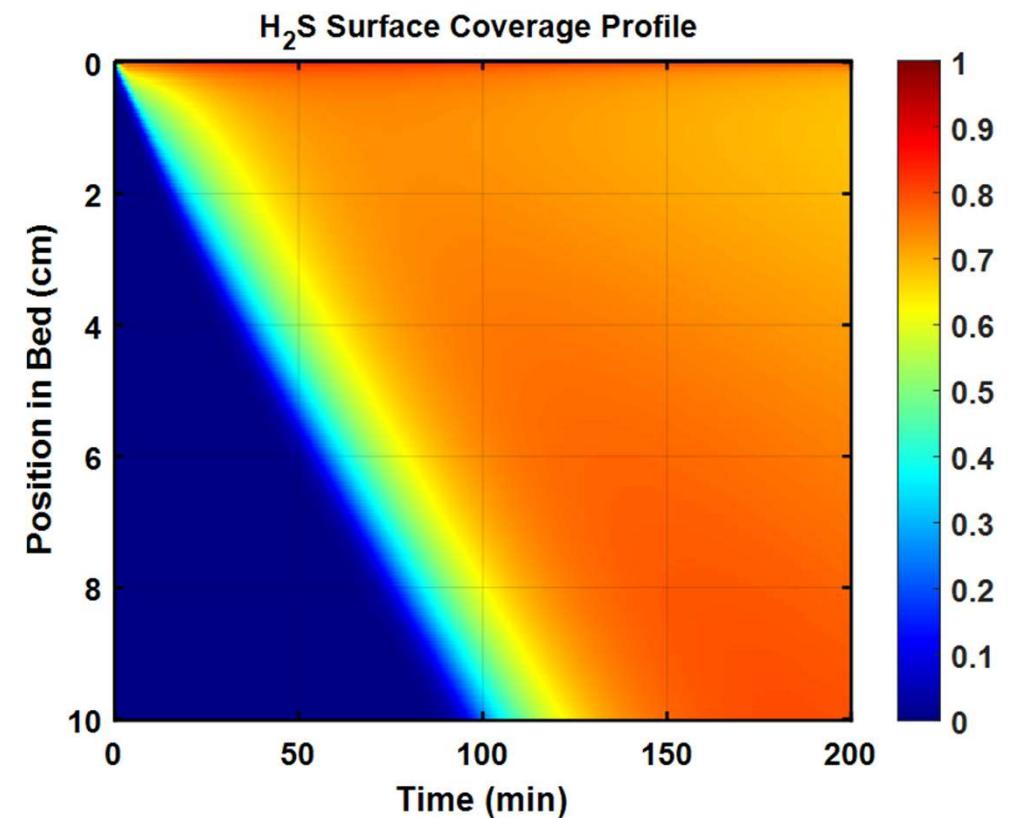
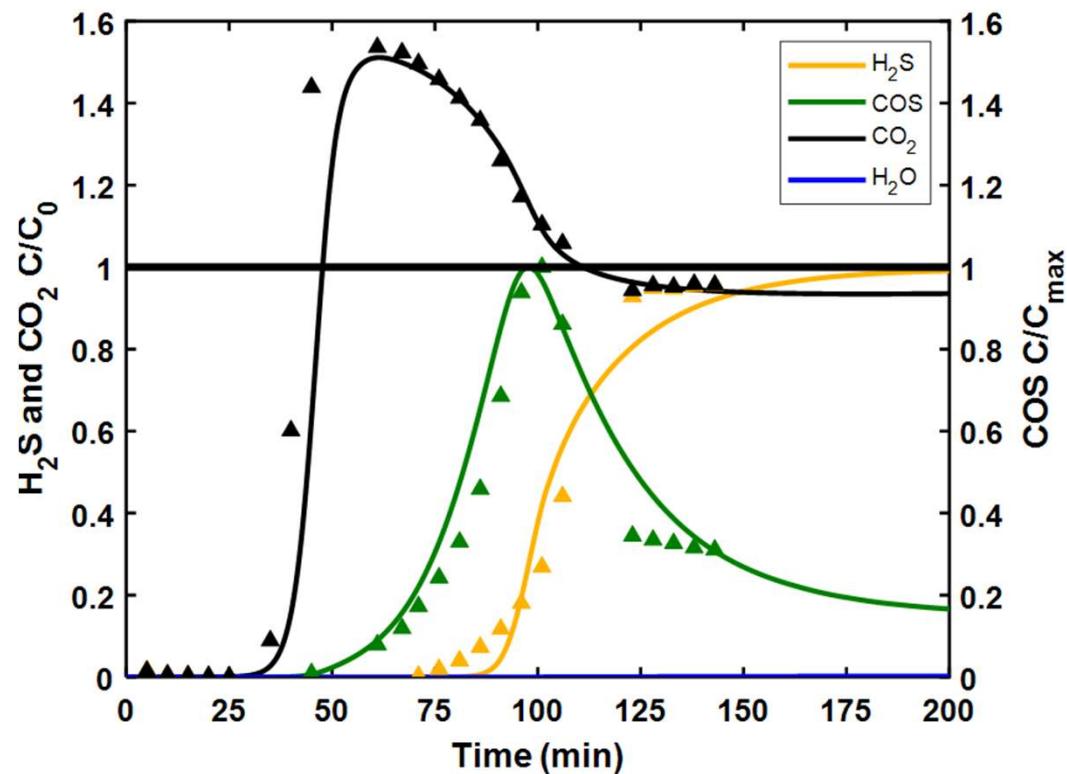
# COS synthesis: experimental



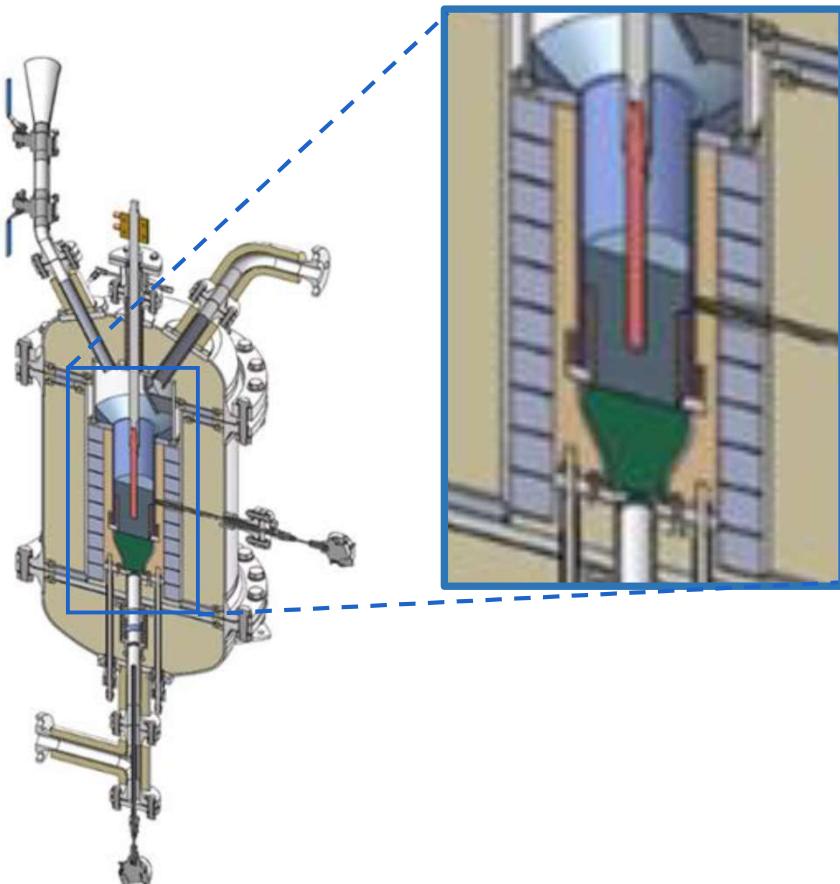
Feed mixture of  $\text{H}_2\text{S}:\text{CO}_2=1:1$  on 13X at  $45^\circ\text{C}$ . Thick line at  $\text{C}/\text{C}_0 = 1$ .



# COS synthesis: modeling



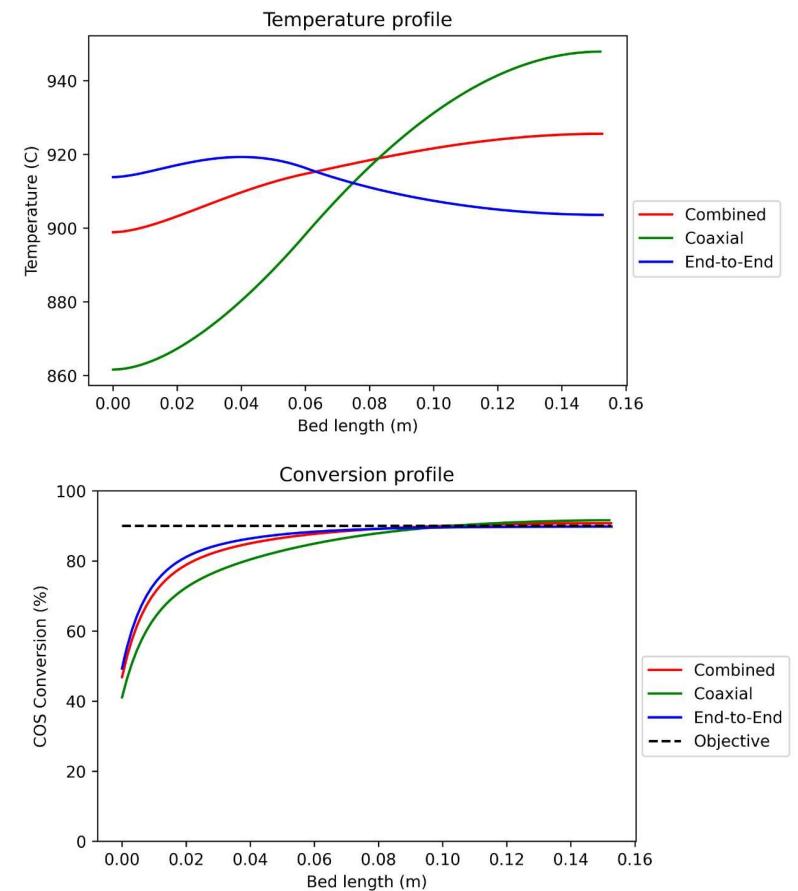
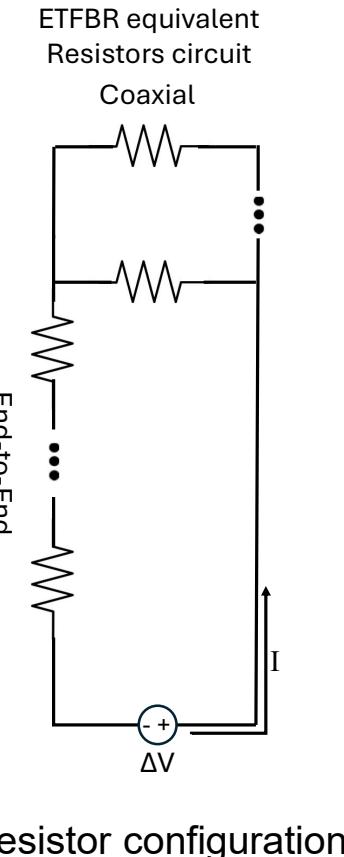
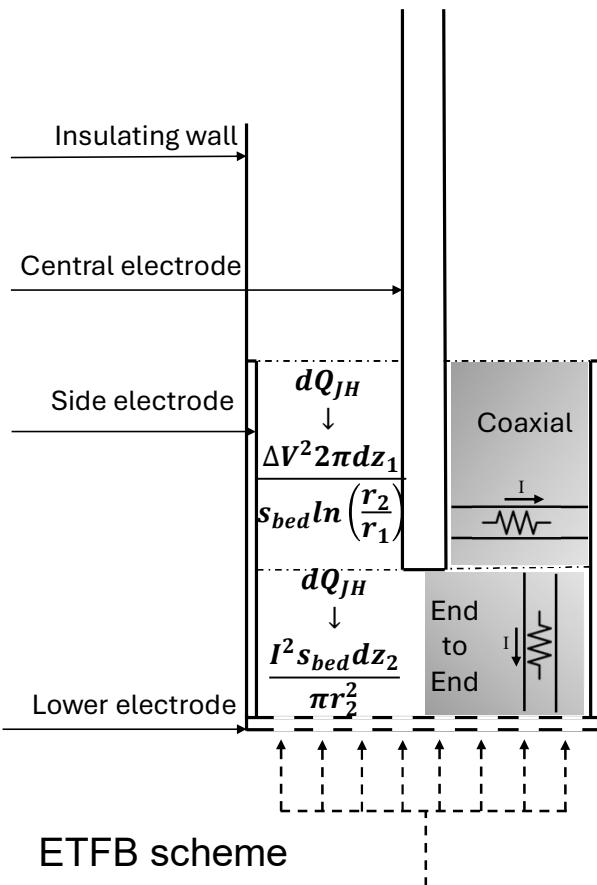
# COS decomposition



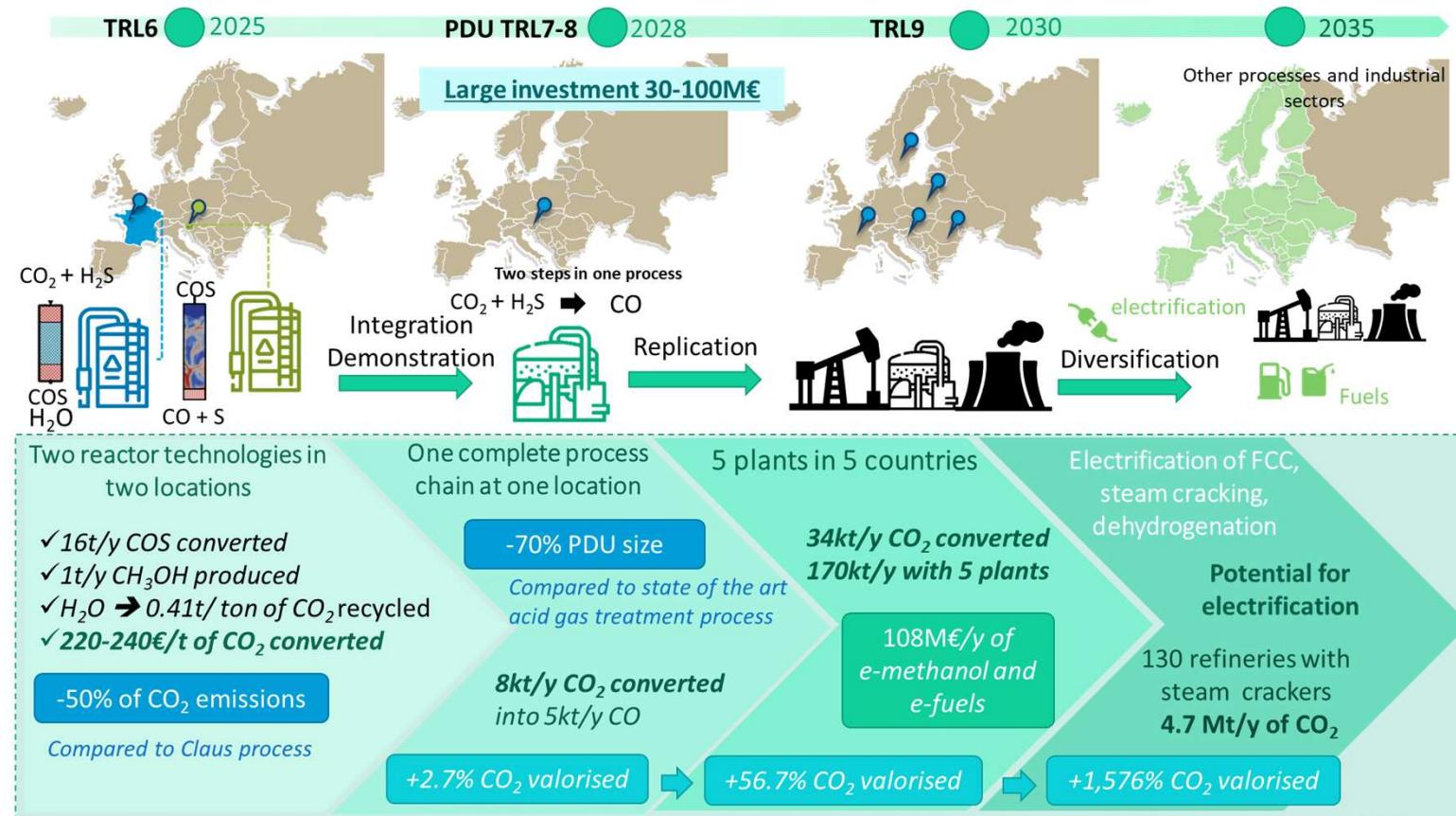
- COS decomposition to CO and sulphur
- temperatures up to 800-1200 °C
- in situ heat generation by joule heating



# e-CODUCT – ETFB modelling



# e-CODUCT valorization roadmap



# conclusions, opportunities and perspectives

- chemical reactor electrification
  - more than connecting an electric heater to the grid
  - reasoning from the inside
  - CO<sub>2</sub> emission reduction
  - integration in a process
  - ...
- challenges
  - few large-scale vs many small-scale applications
  - electricity availability
  - impact on the chemistry
  - ...

# acknowledgments

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# acknowledgements (2)

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Kevin De Ras

**Klaus Jacobs**

Laura Pirro

Loïc Eloi

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# acknowledgements (3)



European Research Council



TotalEnergies



We create chemistry





# 12<sup>th</sup> International Symposium on Catalysis in Multiphase Reactors & 11<sup>th</sup> International Symposium on Multifunctional Reactors

## CONFERENCE THEME

Multiscale modeling and  
experimentation

Reactor design  
Process development  
Low carbon technology  
Renewable chemicals  
Polymer design  
Catalysis and kinetics

8-11 SEPTEMBER 2024

Ghent, Belgium

early bird registration  
deadline: June 30th

MORE INFO

[www.camure.ugent.be](http://www.camure.ugent.be)

## SCIENTIFIC COMMITTEE

**Chairman of the symposium**

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**Co-chairmen of the symposium**

prof. dr. ir. Kevin VAN GEEM

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FACULTY OF ENGINEERING  
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DRIVING CHEMICAL TECHNOLOGY



# e-CODUCT: Want to know more?!



**Website:** <https://e-coduct.eu/>



**LinkedIn:** @e-coduct project  
<https://www.linkedin.com/in/ecoduct/>



**Twitter:** @eCODUCT2022  
<https://twitter.com/eCODUCT2022>



**YouTube:** @ecoduct2022  
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#innovation #technology



## LABORATORY FOR CHEMICAL TECHNOLOGY

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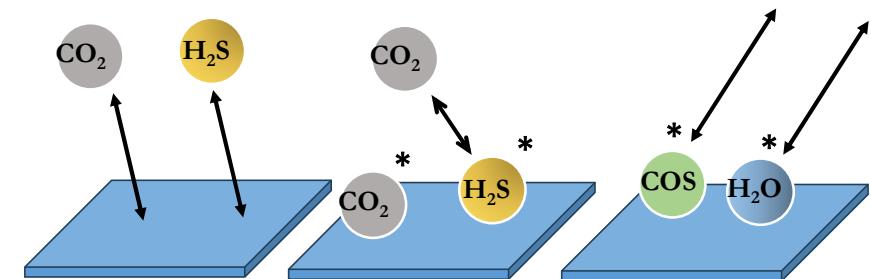
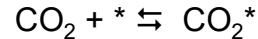
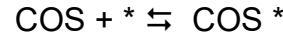
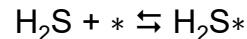
E      [info.lct@ugent.be](mailto:info.lct@ugent.be)  
T      00329331757

<https://www.lct.ugent.be>

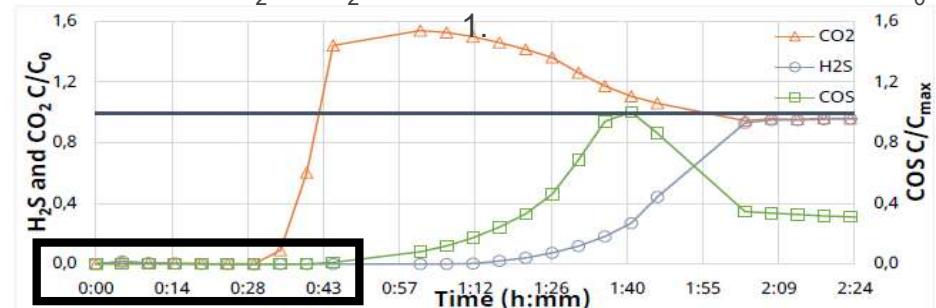


## Modified ER Mechanism

### Eley-Rideal with $\text{CO}_2$ & COS adsorption



Feed mixture of  $\text{H}_2\text{S}:\text{CO}_2=1:1$  on 13X at  $45^\circ\text{C}$ . Thick line at  $C/C_0 =$

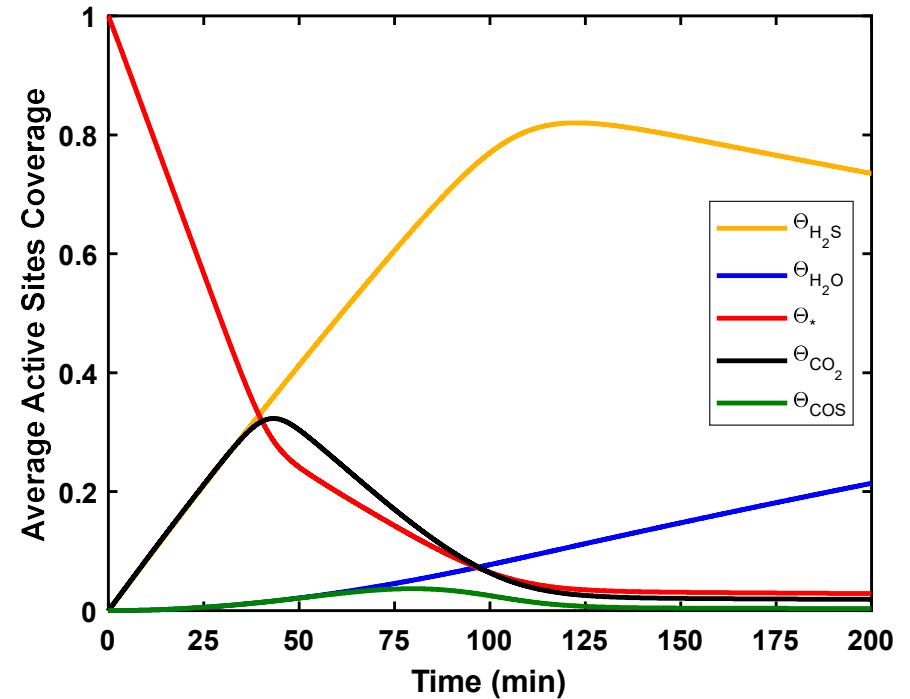
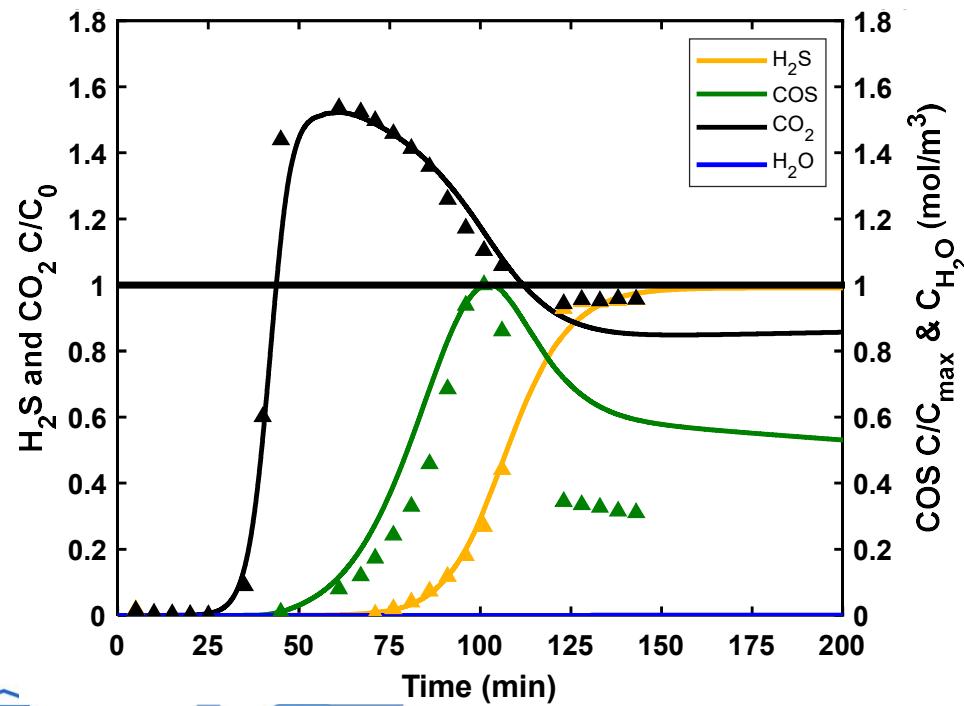


$\text{CO}_2$  and COS retention is accounted

18M General Assembly meeting

## Modified ER Mechanism: Results

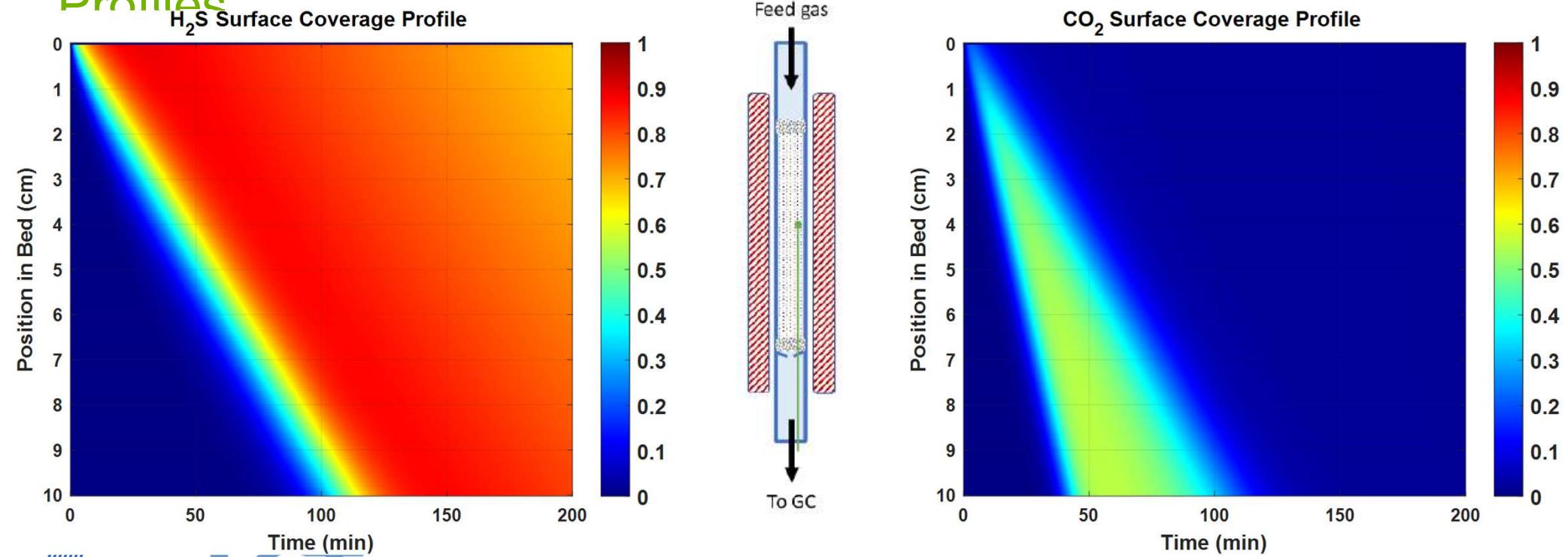
Feed mixture of  $\text{H}_2\text{S}:\text{CO}_2=1:1$  on 13X at 45°C. Thick line at  $\text{C}/\text{C}_0 = 1$ .



	$k_{\text{H}_2\text{S}}^{\text{ads}}$	$k_{\text{H}_2\text{S}}^{\text{des}}$	$k_r^f$	$k_r^r$	$k_{\text{H}_2\text{O}}^{\text{des}}$	$k_{\text{H}_2\text{O}}^{\text{ads}}$	$k_{\text{CO}_2}^{\text{ads}}$	$k_{\text{CO}_2}^{\text{des}}$	$k_{\text{COS}}^{\text{des}}$	$k_{\text{COS}}^{\text{ads}}$
DRIVI	1E-3	1.8E-4	5.5E-6	1E-7	1E-4	1	6E-4	4E-3	3E-3	1E-3

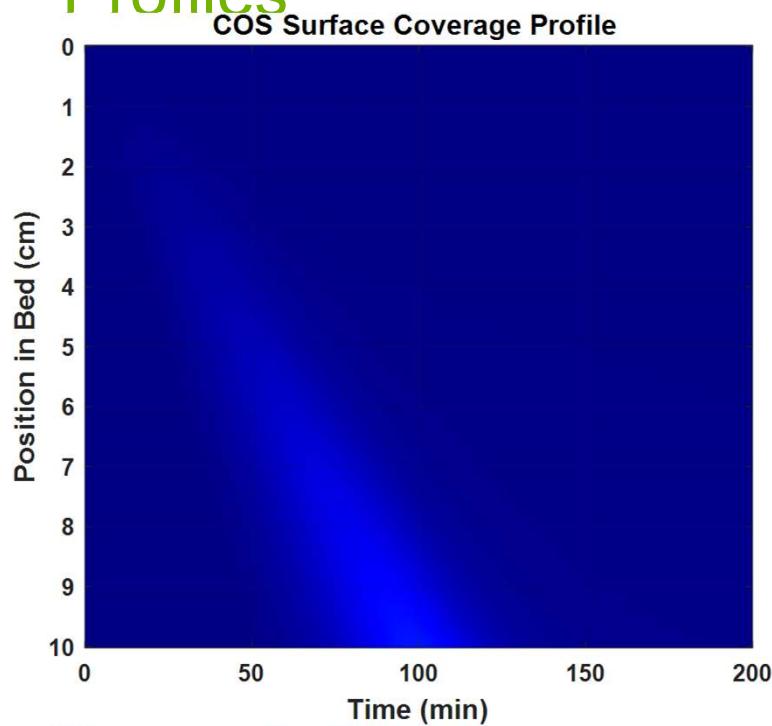
## Modified ER Mechanism: Results – Reactant

### Profiles



## Modified ER Mechanism: Results – Product

### Profiles



Feed mixture of H<sub>2</sub>S:CO<sub>2</sub>=1:1 on 13X at 45°C. Thick line at C/C<sub>0</sub> = 1.

Feed gas

To GC

### H<sub>2</sub>O Surface Coverage Profile

